

Home Search Collections Journals About Contact us My IOPscience

Achieving high mobility ZnO : Al at very high growth rates by dc filtered cathodic arc deposition

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 2011 J. Phys. D: Appl. Phys. 44 232003 (http://iopscience.iop.org/0022-3727/44/23/232003) View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 131.243.223.203 The article was downloaded on 01/08/2011 at 23:42

Please note that terms and conditions apply.

J. Phys. D: Appl. Phys. 44 (2011) 232003 (5pp)

FAST TRACK COMMUNICATION

Achieving high mobility ZnO : Al at very high growth rates by dc filtered cathodic arc deposition

R J Mendelsberg $^{1,2},$ S H N Lim 1, Y K Zhu 3, J Wallig 1, D J Milliron 2 and A Anders 1

¹ Lawrence Berkeley National Laboratory, Plasma Applications Group, Berkeley, CA

² Lawrence Berkeley National Laboratory, Molecular Foundry, Berkeley, CA

³ Harbin Institute of Technology, Harbin, People's Republic of China

E-mail: aanders@lbl.gov

Received 4 April 2011, in final form 7 May 2011 Published 23 May 2011 Online at stacks.iop.org/JPhysD/44/232003

Abstract

Achieving a high growth rate is paramount for making large-area transparent conducting oxide coatings at a low cost. Unfortunately, the quality of thin films grown by most techniques degrades as the growth rate increases. Filtered dc cathodic arc is a lesser known technique which produces a stream of highly ionized plasma, in stark contrast to the neutral atoms produced by standard sputter sources. Ions bring a large amount of potential energy to the growing surface which is in the form of heat, not momentum. By minimizing the distance from cathode to substrate, the high ion flux gives a very high effective growth temperature near the film surface without causing damage from bombardment. The high surface temperature is a direct consequence of the high growth rate and allows for high-quality crystal growth. Using this technique, 500–1300 nm thick and highly transparent ZnO : Al films were grown on glass at rates exceeding 250 nm min⁻¹ while maintaining resistivity below $5 \times 10^{-4} \Omega$ cm with electron mobility as high as $60 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$.

(Some figures in this article are in colour only in the electronic version)

Currently, indium supply can meet the demand, even in the United States where no indium is mined and little is actually recycled. However, building energy efficiency must be improved world wide and smart, multi-functional windows requiring transparent conductive oxide (TCO) coatings will play an important role. About $10^8 \text{ m}^2 \text{ year}^{-1}$ of TCO coated glass is already required for the flat panel display and solar cell industry. Smart windows will require TCO coatings over areas of the same order of magnitude, and the photovoltaic market is also expected to grow. This will require a very substantial increase in the indium supply if indium tin oxide (ITO) and other In based materials remain the TCO of choice. Such an increase in the demand of ITO will have a profound impact on the indium price which is already subject to large swings. Furthermore, indium has very unique properties which make it useful for many other important applications which will inevitably suffer. For example, indium is used for coatings for aircraft parts, for cryogenic and vacuum applications, in optoelectronic devices for fibre-optic communications, etc.

Aluminium-doped ZnO (AZO) is one of the leading candidates to replace ITO but several obstacles must first be overcome. Aside from AZO being less resilient to moisture and acids, the electrical properties of AZO deposited onto glass are not as good as with ITO. Epitaxial AZO grown on sapphire by pulsed laser deposition (PLD) [1] has shown mobility as high as $70 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ but mobility is $40-50 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ when deposited on glass [2, 3]. The resistivity was a record low $0.8 \times 10^{-4} \Omega$ cm in the study by Agura *et al* [2] but PLD has a very low growth rate and is not well suited for large-area deposition.

Magnetron sputtering is the current standard for large-area glass coating by physical vapour deposition. It can produce AZO/glass with resistivity as low as $5 \times 10^{-4} \,\Omega \,\mathrm{cm}$, but typically the mobility is limited to about $30-40 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ [4]. This means a high carrier concentration is present, which limits transmission of the solar infrared that could be used to heat a building in cold climates or help power a solar cell. There is one publication where an exceptionally high mobility of 53 cm² V⁻¹ s⁻¹ was reported for as-deposited rf sputtered AZO which was 780–900 nm thick [5]. However, the growth rate was relatively slow at 15 nm min⁻¹. Reactive dc sputtered AZO has achieved a mobility of $46 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, but no information on the growth rate was reported [6]. Considering that the films were only 300-400 nm thick and that the rate was not even mentioned, it may be reasonable to assume that the growth rate was not considerably higher than that of rf sputtering.

There are a few reports of substantially higher growth rates, reaching 580 nm min^{-1} , for AZO grown by reactive magnetron sputtering but the mobility is limited to $20 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ at best [7]. For AZO, ZnO, and many other crystalline materials, film quality judged by several metrics seems to be inversely proportional to the growth rate of the technique. Atomic layer deposition can produce high quality crystals (of a handful of materials) at very low rates, followed by MBE, and PLD. Then comes RF sputtering, then reactive sputtering and then ultra-high rate techniques like atmospheric plasma torches which are best suited for amorphous materials and thick porous films.

A frequently overlooked growth technique capable of high quality crystal formation is cathodic arc deposition. It is typically disregarded in major review papers on TCOs and has a low profile at most large conferences without a hard coating theme. Pulsed filtered cathodic arcs have produced AZO [8] with mobility above $40 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, but the growth rates are comparable to PLD. However, unlike PLD, cathodic arcs can be operated in a dc mode, giving much higher growth rates. AZO has been produced by dc cathodic arc in the past [9–11], but the quality was inferior to the films produced by reactive magnetron sputtering. This fact is likely why effort into dc arc growth of AZO and other TCOs seems to have diminished.

However, cathodic arc has the inherent advantage of producing a highly ionized plasma of cathode material. Energetic condensation from a plasma is known to give dense films due to both the kinetic and potential energy of the arriving ions [12]. Substrate biasing can control the kinetic energy of the arriving ions, but kinetic energy also brings momentum which can damage the growing film. On the other hand, neutralization of the ion at the film surface releases the potential energy of the films leading to better crystal quality. Thus, it is surprising that dc arc-grown AZO did not previously outperform the sputter-deposited films which condense from neutral atoms with very few ions in the arriving flux of particles.

As we will show, the limiting factor for previous work on dc arc grown AZO is the relatively large path the plasma followed from cathode to substrate. A long path is typically needed to make room for the magnetic plasma filter necessary to produce particulate-free coatings [13], but it significantly reduces the ion flux. In this work, very high quality AZO with mobility as high as $60 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ was produced at rates exceeding 250 nm min⁻¹ by minimizing the distance the plasma travelled. The high quality of these films compared with previous reports is attributed to the relatively high ion flux reaching the surface in the compact arrangement. The high flux locally raises the temperature *at the film surface* and allows for unusually high-quality crystal growth at high rates.

Film deposition was done in a vacuum chamber which reaches a base pressure of 1×10^{-5} Torr when the substrate is at elevated temperatures. Oxygen gas was then fed in at a rate of 20–50 sccm with the total pressure in the range 1–5 mTorr. Optimum oxygen pressure depends strongly on the arc current and these ranges were found to give the best balance for low resistivity and high visible transparency.

Readily available, very-low-cost Zn containing about 4 at% Al was used for the cathode (40\$ kg⁻¹). Arc currents typically used in this system are a relatively modest 25–70 A dc. The arc plasma was filtered using a helical quarter-torus open coil carrying 400 A dc and generating a magnetic field estimated to be 50 mT. Anode–filter separation as well as filter–substrate distance was minimized, with a 300 mm cathode–substrate separation.

Borosilicate microscope glass slides $(25 \text{ mm} \times 75 \text{ mm})$ 1 mm thick) were used as substrates. They were cleaned with Liquinox, a widely used glass detergent containing ethylene diamine tetra acetate, designed to yield completely residue free surfaces. The samples were thoroughly rinsed with tap water and then the substrates were quickly dried using dry nitrogen, leaving a streak-free surface with no visible particulates or residues. Samples were pre-heated to an initial substrate temperature (T_S) of 425 °C at most, but excellent material could be deposited at $T_{\rm S} = 200 \,^{\circ}{\rm C}$, and reasonably well performing material could be deposited on room temperature (RT) substrates. Before deposition the substrates were exposed to a 150 W oxygen plasma from a constricted plasma source [14] for 2 min, as a critical last step in the environmentally friendly cleaning procedure. Films were well adherent, even when grown up to several μ m thick. Based on our personal experience in the lab growing more than 100 samples, this surprisingly simple cleaning procedure gave an initial glass surface equal to or better than those cleaned with solvents (ethanol and acetone) as judged by spurious residues on the surface. Growth time was usually 2 min and limited to 5 min, sometimes by overheating of the substrate if the plasma flux was high.

AZO films grown by the compact dc filtered arc show a high degree of crystallinity and excellent crystallographic alignment of the grains, even for some samples deposited on RT substrates [15]. Al content in the films was measured to be 1.3 ± 0.3 at% using Rutherford backscattering spectroscopy. SEM, AFM and cross sectional TEM images of samples grown at $T_{\rm S} = 200$ °C are shown in figure 1. The (002) planes are very well aligned and the *c*-axis is normal to the substrate [15], except for the first 80–100 nm which TEM shows as crystalline but with small grains and little overall alignment. SEM and AFM would suggest a very small lateral grain size, but the

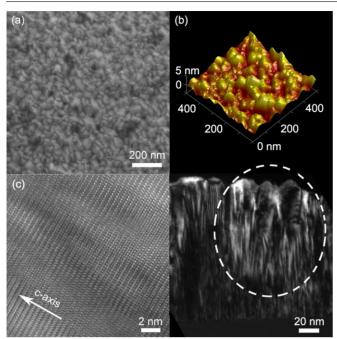


Figure 1. (*a*) SEM, (*b*) AFM, (*c*) cross sectional TEM and (*d*) dark field $(0\,0\,2)$ TEM of AZO grown by dc arc. Images come from three different, typical samples with the TEM images taken on the same sample. A grain is circled in (*d*) with the dotted line.

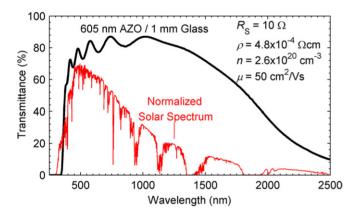


Figure 2. Transmittance of AZO/glass grown by the compact dc arc system (not corrected for the substrate). The AM1.5 solar spectrum is shown in relative units for comparison and the electrical properties of this film are in the upper-right corner.

darkfield (002) TEM image shows the grains to be about 100 nm in diameter in the bulk of the film near the surface.

Optical transmission in the solar spectral range is high in these AZO samples, as shown by one of the better examples in figure 2. For 800–1500 nm photons, this AZO/glass stack shows at least 75% transmittance due to the low carrier concentration. A sheet resistance of $10.0 \pm 0.3 \Omega$ is possible for this 605 ± 15 nm thick sample (resistivity $\rho = 4.8 \times 10^{-4} \Omega \text{ cm}^{-1}$) due to the very high mobility of $50 \pm 3 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. Typical mobility for the samples produced in our system is $40-50 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, with a 1300 nm film showing a record high $60 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, the highest yet reported for as-deposited AZO thin films on glass substrates. For samples deposited at RT, a mobility of $30 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ has been achieved which is much higher than typically reported for RT growth by sputtering [16] and PLD [17]. Properties of the more notable dc arc grown samples are summarized in table 1.

The optical and electrical performance of the films shown in table 1 is noticeably superior to previous reports on dc cathodic arc grown AZO [9–11, 18]. It is very unlikely that the groups who published the reports simply missed the optimized process window. Since we used similar deposition conditions, there must therefore be an important difference in the geometry of the cathodic arc arrangement we used compared with other groups.

Film growth rate (*G*) is sensitive to the geometry of any physical vapour deposition system. For deposition of ZnO using a quarter-torus filter, growth rate is known to linearly increase with increasing arc current [19] (I_{arc}) and linearly decrease with increasing oxygen partial pressure [20] (P_{O_2}). For dc arcs, the growth rate is relatively independent of the substrate temperature (T_S) up to 425 °C [15, 21]. As such, one can approximate the effects of geometry on the ion flux reaching the substrate and the resulting growth rate by multiplying the current-normalized growth rate (G_C) by P_{O_2} .

As shown in table 2, our G_C and $G_C \times P_{O_2}$ values are significantly higher than the others, proving that the much larger ion flux reaching the substrate (and higher G_C) is due to the geometry of our system. Published schematic drawings [20–22], and pictures [23] suggest most other groups have a longer path from cathode to substrate due to the large filters (by an estimated factor of around 2). However, any extra distance the plasma must travel through the O₂ gas reduces its particle and energy flux density by scattering and reduction of the mean ion charge state. Thus, a much greater percentage of the available energy of the cathodic arc plasma is utilized for film growth in the compact arrangement.

The large amount of available energy from the compact arc is critical for obtaining high mobility AZO. Atomic scale heating of the film surface leads to a *growth* temperature which can be higher than the *substrate* temperature [12]. During deposition, the heat from the surface of the growing film can warm the substrate by up to 200 °C, but most of the energy is dissipated on ps time and nm length scales [24]. Therefore, in order to effectively heat the film surface, the ion flux must be high enough to ensure the individual thermal spikes from each ion are close in space and time.

In this compact arrangement the *growth* temperature may even exceed the melting point of the substrate without causing damage. AZO has a much higher melting point than glass, and the high *growth* temperatures reached by the compact arc promote the formation of high-quality crystalline films with large grains. Such a situation cannot be attained by conventional substrate heating. This perhaps greatest advantage of cathodic arc deposition is substantially diminished as the plasma path length increases. Simply increasing the bias voltage on the substrate will lead to film damage and poor electrical performance due to the high kinetic energy of the arriving ions.

In summary, AZO thin films with mobilities in the $40-60 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ range were deposited on glass substrates at rates exceeding 250 nm min^{-1} using a compact dc cathodic arc arrangement. The short plasma path was evidently the

Table 1. Properties of the more notable AZO samples produced by the compact dc cathodic arc system. T_{sol} is the transmittance over the entire AM1.5 spectrum shown in figure 2 through the AZO/glass stack. R_q is the RMS surface roughness.

-	-	-		-	1		-
Sample	d (nm)	R _q (nm)	T_{sol} (%)	$R_{\rm S}$ (Ω)	ρ (10 ⁻⁴ Ω cm)	$n (10^{20} \mathrm{cm}^{-3})$	$\mu (cm^2 V^{-1} s^{-1})$
Highest μ	1290	6.9	75	3.3	3.9	2.7	60
Lowest ρ	505	3.7	77	6.8	3.0	3.8	55
Smoothest	34	0.15	92	397	18.4	1.7	21
$d < 100 \rm{nm}$	95	0.32	88	61	6.0	2.8	38
$T_{\rm s} = 23 ^{\circ}{\rm C}$	310	1.9	78	52	15.0	0.1	30

Table 2. Comparison of dc filtered (quarter-torus) cathodic arc growth rates of ZnO and AZO. ZnO reports are shown above the line, AZO below. I_{arc} is the arc current (A), P_{O_2} is the oxygen pressure (mTorr), *G* is the film growth rate (nm min⁻¹), and G_C is the current-normalized growth rate (nm A⁻¹ min⁻¹).

Report	$I_{\rm arc}$	P_{O_2}	G	$G_{\rm C}$	$G_{\rm C} \times P_{\rm O_2}$
Xu [21]	70	1	20	0.3	0.3
Wang [22]	60	0.1	110	1.83	0.2
-	60	1	30	0.5	0.5
David [<mark>19</mark>]	100	3	127	1.27	3.8
	300	3	650	2.2	6.5
Zhitomirsky [11]	200	5	130	0.65	3.3
This work	30	5	100 ^a	3.3	17
	70	5	300 ^a	4.3	21

^a Average of many samples.

key to maximizing the kinetic and potential energy flux of the condensing plasma, allowing significant transient heating of the growing film surface without overly heating the substrate. The low sheet resistance, high VIS–NIR transparency, and high growth rate of AZO grown by a compact cathodic arc arrangement could prove very competitive with commercially available TCOs used for large-area applications.

Acknowledgments

The authors would like to thank K M Yu and J M Kennedy. Research was supported by the LDRD Program of Lawrence Berkeley National Laboratory, by the Assistant Secretary for Energy Efficiency and Renewable Energy, Office of Building Technologies, and by an Office of Basic Energy Sciences Early Career Research Program grant (Dr Milliron) under US Department of Energy Contract No. DE-AC02-05CH11231. Portions of this work were performed as a User project at the LBNL Molecular Foundry, which is supported by the Office of Science, Office of Basic Energy Sciences, under the same contract.

References

- Lorenz M *et al* 2003 Optical and electrical properties of epitaxial (Mg,Cd)_xZn_{1-x}O, ZNO, and ZNO : (Ga,Al) thin films on *c*-plane sapphire grown by pulsed laser deposition *Solid-State Electron.* 47 2205–9
- [2] Agura H, Suzuki A, Matsushita T, Aoki T and Okuda M 2003 Low resistivity transparent conducting Al-doped ZnO films prepared by pulsed laser deposition *Thin Solid Films* 445 263–7

- [3] Tanaka H, Ihara K, Miyata T, Sato H and Minami T 2004 Low resistivity polycrystalline ZnO: Al thin films prepared by pulsed laser deposition J. Vac. Sci. Technol. A 22 1757–62
- [4] Liu H Y, Avrutin V, Izyumskaya N, Ozgur U and Morkoc H 2010 Transparent conducting oxides for electrode applications in light emitting and absorbing devices *Superlatt. Microstruct.* 48 458–84
- [5] Zhu H, Hupkes J, Bunte E, Gerber A and Huang S M 2010 Influence of working pressure on ZnO: Al films from tube targets for silicon thin film solar cells *Thin Solid Films* 518 4997–5002
- [6] Cornelius S, Vinnichenko M, Shevchenko N, Rogozin A, Kolitsch A and Moller W 2009 Achieving high free electron mobility in ZnO: Al thin films grown by reactive pulsed magnetron sputtering *Appl. Phys. Lett.* **94** 042103
- [7] Hirohata K, Nishi Y, Tsukamoto N, Oka N, Sato Y, Yamamoto I and Shigesato Y 2010 Al-doped ZnO (AZO) films deposited by reactive sputtering with unipolar-pulsing and plasma-emission control systems *Thin Solid Films* 518 2980–3
- [8] Anders A, Lim S H N, Yu K M, Andersson J, Rosen J, McFarland M and Brown J 2010 High quality ZnO : Al transparent conducting oxide films synthesized by pulsed filtered cathodic arc deposition *Thin Solid Films* 518 3313–9
- [9] Goldsmith S 2006 Filtered vacuum arc deposition of undoped and doped ZnO thin films: Electrical, optical, and structural properties *Surf. Coat. Technol.* 201 3993–9
- [10] Takikawa H, Kimura K, Miyano R and Sakakibara T 2002 Cathodic arc deposition with activated anode (cadaa) for preparation of *in situ* doped thin solid films *Vacuum* 65 433–8
- [11] Zhitomirsky V N, Cetinorgu E, Adler E, Rosenberg Y, Boxman R L and Goldsmith S 2006 Filtered vacuum arc deposition of transparent conducting Al-doped ZnO films *Thin Solid Films* 515 885–90
- [12] Anders A 2002 Atomic scale heating in cathodic arc plasma deposition Appl. Phys. Lett. 80 1100–2
- [13] Anders A 2008 Cathodic Arcs: From Fractal Spots to Energetic Condenstion (New York: Springer)
- [14] Anders A and Kuhn M 1998 Characterization of a low-energy constricted-plasma source *Rev. Sci. Instrum.* 69 1340–3
- [15] Mendelsberg R J, Lim S H N, Milliron D J and Anders A 2011 High rate deposition of high quality ZnO: Al by filtered cathodic arc MRS Proc. submitted
- [16] Wang Y, Lu J, Bie X, L Gong, Li X, D Song, Zhao X, Ye W and Ye Z 2011 Transparent conductive Al-doped ZnO thin films grown at room temperature *J. Vac. Sci. Technol.* A 29 031505
- [17] Sakai N, Umeda Y, Mitsugi F and Ikegami T 2008 Characterization of zinc oxide thin films prepared by pulsed laser deposition at room temperature *Surf. Coat. Technol.* 202 5467–70
- [18] Lee H W, Lau S P, Wang Y G, Tse K Y, Hng H H and Tay B K 2004 Structural, electrical and optical properties of Al-doped ZnO thin films prepared by filtered cathodic vacuum arc technique J. Cryst. Growth 268 596–601

- [19] David T, Goldsmith S and Boxman R L 2005 Dependence of zinc oxide thin film properties on filtered vacuum arc deposition parameters J. Phys. D: Appl. Phys. 38 2407–16
- [20] David T, Goldsmith S and Boxman R L 2004 Electro-optical and structural properties of thin zno films, prepared by filtered vacuum arc deposition *Thin Solid Films* 447-448 61-7
- [21] Xu X L, Lau S P and Tay B K 2001 Structural and optical properties of ZnO thin films produced by filtered cathodic vacuum arc *Thin Solid Films* 398–399 244–9
- [22] Wang Y G, Lau S P, Lee H W, Yu S F, Tay B K, Zhang X H, Tse K Y and Hng H H 2003 Comprehensive study of ZnO films prepared by filtered cathodic vacuum arc at room temperature J. Appl. Phys. 94 1597–604
- [23] Tay B K, Zhao Z W and Chua D H C 2006 Review of metal oxide films deposited by filtered cathodic vacuum arc technique *Mater. Sci. Eng.* R 52 1–48
- [24] Hofsäss H, Feldermann H, Merk R, Sebastian M and Ronning C 1998 Cylindrical spike model for the formation of diamondlike thin films by ion deposition *Appl. Phys.* A 66 153–81